TOLGYESSY, G

HUNGARY/Discuses of Farm Animals - Discuses Caused by Bacteria R-2

and Fungi.

Abs Jour : Ref Zhur - Biol., No 14, 1958, 64644

Author : Tolgyesi, Gyorgy

Inst
Title : Study of Proteins by the Method of Electrophoresis on

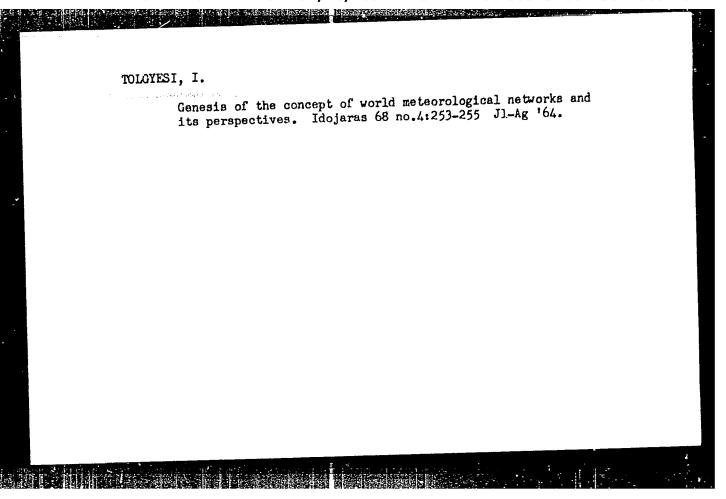
Paper in Veterinary Medinine. I. Leptospirosis and

Tetanus.

Orig Pub : Magyar allatorv. lapja, 12, No 10, 305-309

Abstract : No abstract.

Card 1/1



TOLGYESSY, Gyorgy

ų,

Radiometric titration with marked ferrocyanide volumetric solution.
Magy kem folyoir 65 no.4:149-152 Ap 34.

1. Oddelenie radiochemie pri Katedre fyzikalnej chemie Slovenskej vysokej skoly technickej, Bratislava, CSR.

HUNGARY

HARASZTI, Ede, Dr. Assistant Professor (egyetemi adjunktus), Department of Animal Hygiene (Allathigieniai Tanszek) (Chairman: KOVACS, Ferenc, Dr. Associate Professor (egyetemi docens), vandidate of veterinary medicine) and Department and Clinic of Internal Medicine (Belgyogyaszati Tanszek es Klinika) (Chairman: HORVATH, Zoltan, Dr. Associate Professor, candidate of veterinary medicine) of the University of Veterinary Medicine (Allatorvostudomanyi Egyetem).

TOLGYESI, Gyorgy, Assistant Professor.

"Molybdenum Content of Domestic Grass Species."

Budapest, Magyar Allatorvosok Lapja, Vol 17, No 11, Nov 62, pp 417-419.

Abstract: [Authors' English summary] The molybdenum content of 75 grass species was determined with a colorimetric method. The results are tabulated. They show that the molybdenum content of Hungarian grass species does not rise above the limits of toxicity. Small areas may exist, however, where the soil conditions may favor the absorption of molyklenum. The authors propose to identify such areas. [Seven Hungarian, 6 Western, 2 Fast German references.]

ToloyEssy, Gyorgy

HUNGARY/Analytical Chemistry - Analysis of Inorganic Substances.

: Ref Zhur - Khimiya, No 8, 1958, 24728 Abs Jour

Tolgyessy Gyorgy, Schiller Pal Author

Determination of Ag-, Pd2+ and Cu+ Ions by the Method of Radiometric Titration Using I131 Indicator. Inst Title

: Magyar kem. folyoirat, 1957, 63, No 10, 269-271 Orig Pub

: Description of a method of titrimetric determination of Abstract

Ag+, Pd2+ and Cu+, forming a precipitate with I-; the titration end point is determined by measuring the activity of the solution over the resulting precipitate (to 200 ml of 0.1 N solution of KI is added 1 ml of a solution of isotope Il31 having an activity of 0.842 mcurie/ml). To measure the activity use is made of an apparatus which makes it possible to draw the solution, through a tube, from the titration flask into a spiral tube surrounding a Geiger-Mueller counter. Equivalence points are determined

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HUNGARY/Analytical Chemistry - Analysis of Inorganic Substances.

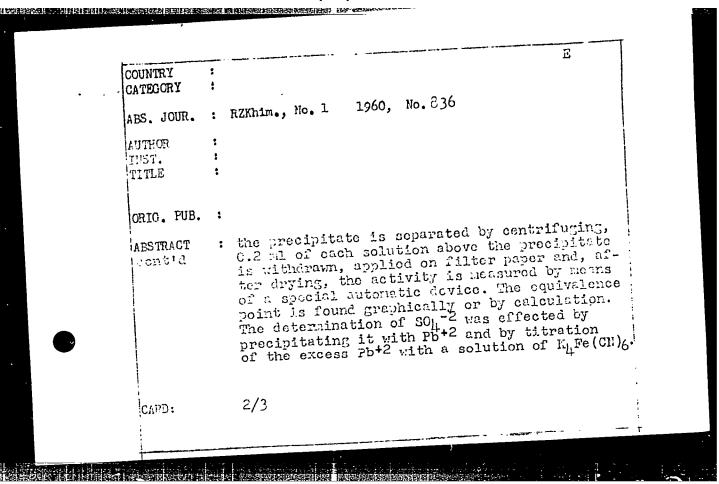
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Abs Jour : Ref Zhur - Khimiya, No 8, 1958, 24728

by calculation, or graphically, making a correction for change in volume. The graphic method is considerably faster and more accurate. Duration of one determination is 8-10 minutes; the error is of about 1%.

Card 2/2



COUNTRY CATTMORY: ABS. JOUR. : RZKhim., Mo. 1 1960, No.336 AUTHOR: INST. : TITLE: ORIG. PUB.: ABSTRACT: In the joint presence of Pb+2 and Zn+2, first the sum Pb+2 + Zn+2 is titrated, then pb+2 is precipitated with sulfuric acid and Zn+2 is titrated I. Krishtofori			E	
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TOLGYESSY, GY.

Radiometric titration by marked potassium ferrocyanide used as volumetric solution. p. 149.

MAGYAR KEMIAI FOLYOIRAT. Budapest, Hungary. Vol. 65, no. 4, Apr. 1959

Monthly List of East European Accessions (EEAI), LC. Vol. 8, No. 9, September 1959 Uncl.

TOLGYESSY, Juraj, doc. inz. CSc.; POPOV, Christomil Petkov; STEFANOV, Georgi Ivanov; TOMOV, Trifon Tomov, inz.

THE REPORT OF THE PARTY OF THE

Nondestructive determination of indium in intermetallic alloys by neuron activation in using Fo+Be neutron source.Chem zvesti 18 no.1:48-55 164

1. Nauchno izsledovatelski geologicheski institut pri Glavno upravlenie po geologiia i okhrana na zemnite nedra, Laboratoriia akticatsionen analiz, Sofiia (for all except Togyessy).

2. Katedra radiochemie a radiacnej chemie, Slovenska vysoka skola technicka, Bratislava (for Tolgyessy).

TOLGYESSY, Juraj, doc. inz., CSc.; HRADIL, Miroslav; JESENAX, Viktor, doc. inz., CSc.; BRAUN, Tibor, dr.

Rediocoulometric titration by using nonisotopic solid-phase indicators. Chem zvesti 19 no.6:465-469 '65.

1. Chair of Radiochemistry and Radiation Chemistry of the Slovak Higher School of Technology, Bratislava, Janska ulica (for Tolgyessy and Hradil). 2. Chair of Inorganic Technology of the Slovak Higher School of Technology, Bratislava, Janska ulica (for Jesenak). 3. Chair of Inorganic and Analytic Chemistry of Lorand Eotvos University, Budapest, VIII., Muzeum korut 4/b (for Braun). Submitted December 10, 1964.

TOLOWESTY, Juraj, doc. ins. Call.

"Nuclear techniques in analytical chemistry by a.J. Moses. Reviewed by J. Polgyessy. Them 200 att 19 no.b; 604-507 (65.

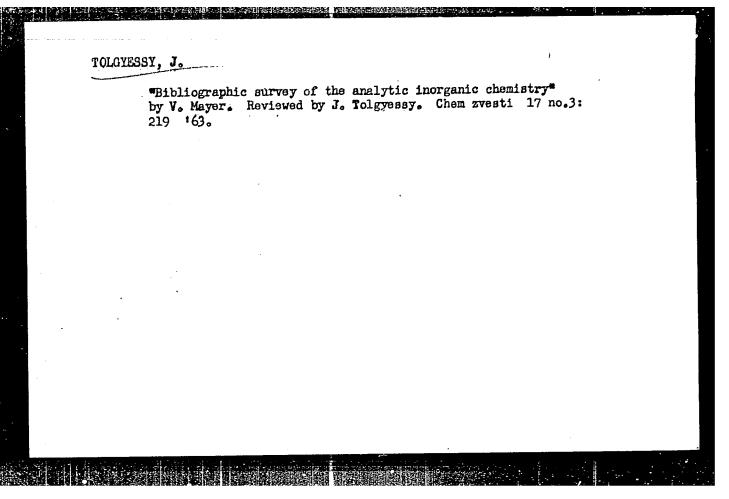
EN LA TIME STATE STATE OF THE BEST OF THE STATE OF THE ST

1. Chair of Radiochemistry and Radiation Chemistry of the Sicvak Higher School of Technology, Bratisians, Janska ulica, and Editorial Board Member, "Chemiske svesti."

TOLGYESSY, J., KRIVAH, V.

Possibilities of application of radioisotopes and their use in Slovakia; p. 535 TECHNICKA PRACA. Czechoslovakia, Vol. 11, No. 7, July 1959

Monthly List of East European Accessions (EEAI), LC. Vol. 8, No. 9, Sep 1959 Uncl.



TOLGYESSY, J.

"Technique of inorganic chemistry", edited by N.R. Johnson, E. Eichler, G.D. O'Kelley. Reviewed by J. Tolgyessy. Chem zvesti 18 no.11:879-880 '64.

1. Editorial board member, "Chemicke zvesti."

CZECHOSLOVAKIA

J. TÖLGYESSY and M. SARSUNOVA, Chair of Radiochemistry and Radiation Chemistry of the Faculty of Chemistry (Katedra radiochemie a radiacnej chemie Chemickej fakulty) SVST [Slovenska vysoka skola technicka = Slovak College of Engineering], and KKL [Abbr. not identified] - KUMZ [Krajsky ustav narodniho zdravi, Kraj Institute of National Health] of Western Slovak Kraj (zapadoslovenskeho kraja,) Bratislava.

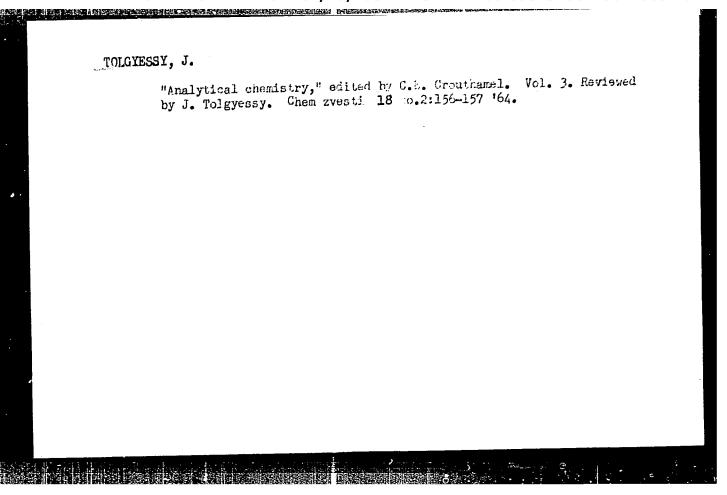
"Use of Isotope Dilution Method in Analytical Biochemistry and Drug Control."

Prague, Ceskoslovenska Farmacie, Vol 12, No 5, June 63; pr 257-262.

Abstract: A general review of the basic mathematical principles underlying methods for using isotopes with analytical purposes in pharmacomedical sciences. Many mathematical formulae; 91 Western, 10 Soviet and 4 Czech references.

11/1

30



TOLGYESSY, Juraj; KRIVAN, Viliam; VARCA, Stefan; KIAS, Jan

Utilization of the beta ray reflection in chemical analysis. Jaderna energie 10 no. $3:^{95}$ Mr $^{1}64.$

THE PROPERTY OF THE PARTY OF TH

1. Department of Radiocher is by and Radiation Chemistry, Faculty of Chemistry, Slow Higher School of Technology, Bratislava.

TOLGYESSY, Juraj; DILINGER, Pavel

Beta absorption radiometric precipitation titrations. Jaderna energie 10 no. 3:86 Mr 164.

1. Department of Radiochemistry and Radiation Chemistry, Faculty of Chemistry, Slovak Higher School of Technology, Bratislava.

TOLGYESSY, Juraj, doc., inz., C.Sc.; KLAS, Jan, promovany chemik; SARSUNOVA, Magda, dr., PhMr., C.Sc.

Determining the calcium salts by measuring the intensity of reflected beta radiation. Chem zvesti 17 no.2:140-145 '63.

1. Katedra radiochemie a radiacnej chemie, Slovenska vysokaskola technicka, Bratislava, Kollarovo namesti 2 (for Tolgyessy and Klas). 2. Krajske kontrolne laboratorium, Krajsky ustav narodneho zdravie Zapadoslovenskeho kraja, Bratislava, Radlinskeho 20 (for Sarsunova).

ICLGYES

CZECHOSLOVIKIA /Physical Chomistry. Radiation Chemistry, Photochomistry, Theory of Photographic Process.

B-10

Abs Jur

: Rof Zhur - Khim., No 10, 1958, No 31834

Luthor

: J. Tolgyessy, P. Kovacs.

Inst

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: Upon the U:e of Thermochromatic Properties of ${\rm lig}_2/{\rm HgI}_4/{\rm lig}_4$ for Supersonic Detection.

Orig Pub

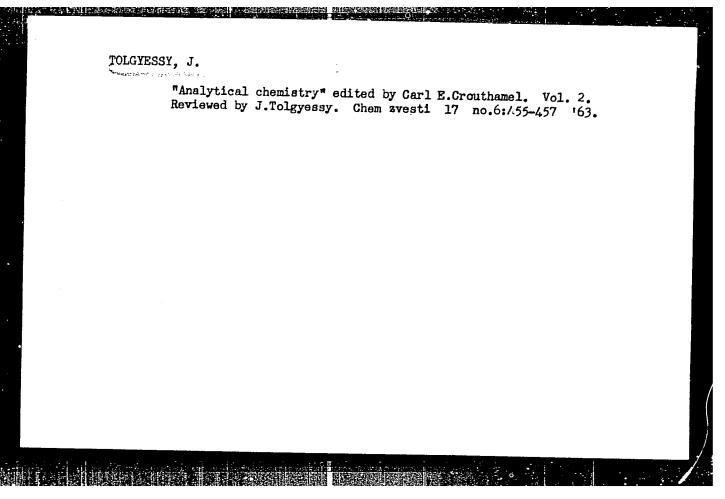
: Chom. zvesti, 1955, 7, No 5, 306-309.

Libstract

: Now mothods of preparing supersonic indicators based on the change of Ag2(HgI4) color are described. The best results were obtained by the application of methylmethacrylate as a dispersion medium on an acetylcellulose film.

Card 1/1

24



TOLGYESSY, Juraj, doc., inz., CSc.; VAIGA, Stefan, doc., inz., CSc.

Use of beta ray reflections for indication of the equivalence point in precipitation titration determination. Chem zvesti 17 no.10/11:779-786 163.

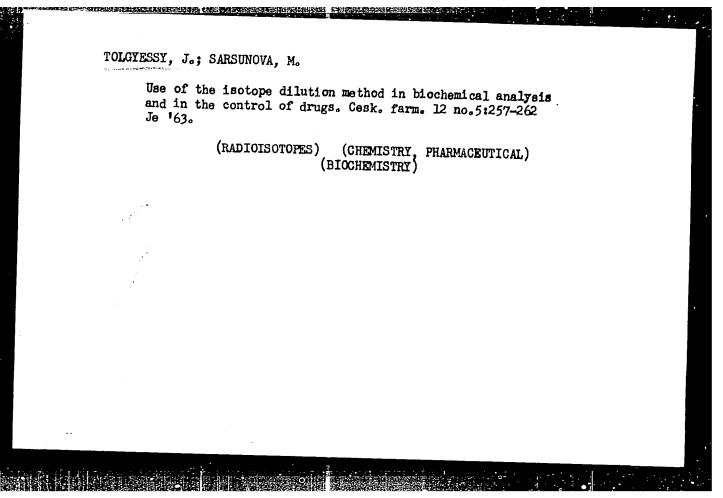
1. Katedra radiochemie a radiacnej chemie, Slovenska vysoka skola technicka, Bratislava, Kollarovo namesti 2.

TOLGYESSY, Juraj, doc., inz., CSc.; DILLINGER, Pavol, promovany chemik

A THE RESIDENCE OF THE PROPERTY OF THE PROPERT

Examination of the use of beta ray absorption for determining uranium and thorium in aqueous solutions. Chem zvesti 17 no.6: 439-444 '63.

1. Katedra radiochemie a radiacnej chemie, Slovenska vysoka skola technicka, Bratislava, Kollarovo namesti 2.



2/043/63/000/002/002/003

AUTHORS:

Tolgyessay, J., Šaršúnová, H., Klas, J.

TITLE:

Determination of calcium salts by measuring the intensity of reflected

beta radiation.

Chemické zvesti, no. 2, 1963, 140-145. PERIODICAL:

TEXT: Beta radiation particles change their direction of flow by interaction with atoms. Backward flow is called reflected radiation. Mechanism of this is complicated and not fully explained; empirical knowledge allows use of it in chemical analysis. Reflection depends on atomic number of the examined material, thickness of the sample, its geometrical shape and physical properties. The authors applied the method for the determination of Ca in an injection solution of Ca gluconate, CaCl2 solution and in yellow sulfur ointment (Unguentum sulfuratum flavum). The apparatus consisted of a ring shaped radiation source with a lead shield and a Geiger-Muller counter. The authors used Tl 204 with a half-life of 2.7 years and an energy of 0.76 MeV. A working method is described and a calibration curve is shown, which was designed for the apparatus using samples of known concentration of calcium. Results obtained with the 3 chemicals mentioned were within 1%. The method gives fast results without damage to the analyzed sample. 3 figures, 3 tables, 2 Western, 6 Czech

Card 1 of 2

Z/043/63/909/002/002/002/003

Determination of calcium....

references.

ASSOCIATION: Katedra radiochémie a radiavnej chémie Slovenskej vysokej školy technickej, (Chair of Radiochemistry and Nuclear Chemistry at the Slovak Technical University) Bratislava. Krajské kontrolné laboratorium KUNZ Západoslovenského kraja (Regional Control Laboratory of the Public Health Institute of Western Slovakia) Bratislava.

Card 2 of 2

TOLGYPSSY, J.; SARSUNOVA, M., dr. (Bratislava, Vazovova 34); HAIS, I.M.

Use of radioisotopes in some chromatographic methods and their application in pharmacy and drug research. Cesk. farm. 14 no.1:9-21 Ja 165

1. Katedra radiochemie a radiacnej chemie Slovenskej vysokej skoly technickej, Bratislava; Krajske kontrolne laboratorium Krajskeho ustavu narodniho zdravi Zapadoslovenskeho kraja, Bratislava i Ustav pro lekarskou chemii lekarske fakulty Karlovy University, Hradec Kralove.

TOLGYESSY, J.; JESZENAK, V.

Automatic thermometric titration. p. 385. CHEMICKE ZVESTI. Bratislava. Vol. 9, no. 6, June 1955.

1st National Conference on Inorganic Chemistry. p. 390.

SOURCE: East European Accessions List (EEAL), LC, Vol. 5, no. 3, March 1956

TOLGYESSY, J.; KOVACS, P.

Searching for deficiencies in metallic objects using Geiger-Muller counters with radioactive elements, $\ p.\ 11.$

Vol. 8, no. 1, Jan. 1956 TECHNICKA PRACA Bratislava, Czechoslovakia

Source: East European Accession List. Library of Congress Vol. 5, No. 8, August 1956

TOLGYESSY, J.

8(1,2)

PHASE I BOOK EXPLOITATION

SLOV/2498

Berčík, Juraj, Engineer and Juraj Tölgyessy, Engineer

Potenciometria (Potentiometric Analysis) Bratislava, SVTL, 1957. 327 p. 1,700 copies printed.

Reviewers: Samuel Stankoviansky, Engineer and Gabriel Dušinsky, Engineer; Resp. Ed.: Jozef Mjartan; Tech. Ed.: F.R. Blažko; Managing Ed. for Literature on Chemistry: Pavol Holéczy.

PURPOSE: This book is intended for research analysts of plants and research organizations. It may also be useful to biologists, physicians and students of higher and secondary schools specializing in chemistry.

COVERAGE: The authors present fundamentals of electrical engineering and discuss the theory of electrolytes and methods of measuring emf. They also discuss potenticmetric titration and describe methods of measuring and automatic recording and control of pH. The authors thank Professor D. Pristavk, Engineer and Chairman of the Chair of Analytical Chemistry of SVST in Bratislava for his help in preparing the text. They thank I. Mayer, Engineer

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Potentiometric Analysis

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and Assistant to the Chairman of the Chair of Theoretical and Experimental Electrical Engineering of SVST, Bratislava, for his help in writing Chapter I. They also thank J. Drobny, Engineer, for providing photographs and graphs and G. Dušinský, Engineer, for reviewing the manuscript. There are 194 references: 47 Czech, 28 Soviet, 57 English, 48 German, 3 Polish, 4 Hungarian, 2 Danish, 2 French, 1 Dutch, 1 Italian and 1 Rumanian.

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TOLGYESSY, J.

CZECHOSLOVAKIA / Chemical Technology, Chemical Products and Their H-25

Application. Fats and Oils. Waxes. Scap and

Detergents. Flotation Agents.

Abs Jour : Ref Zhur - Khimiya, No 5, 1959, No. 17146

Author : Tolgyessy, J.; Robinson, T.

Inst : Not given

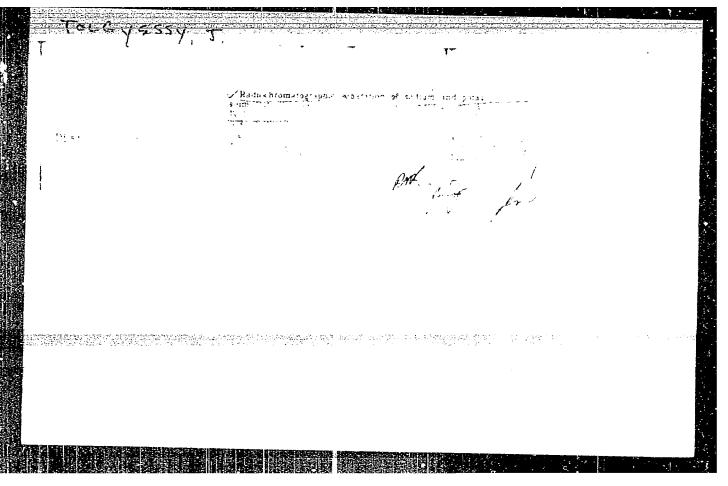
Title : Evaluation of Detergency by Means of Artificial Polution

with Radioactive Isotopes

Orig Pub : Textil (Ceskosl.), 1957, 12, No 11, 414-418

Abstract : No abstract given

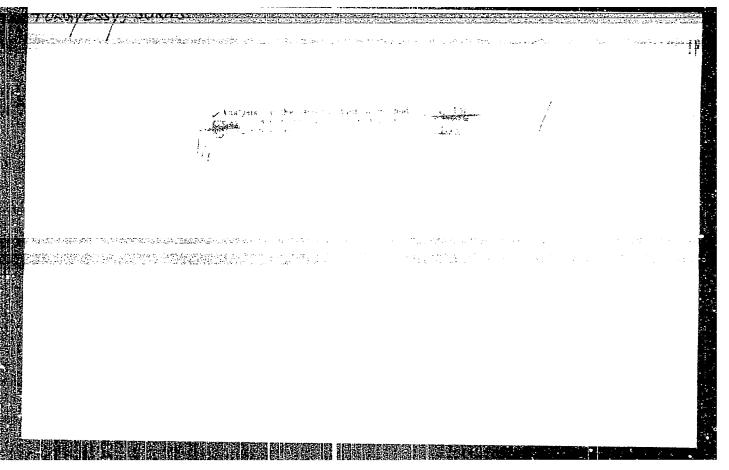
Card 1/1



TOLGYESSY, Juraj, inz., C.Sc. (Bratislava, Kollarovo nam. 2, Chemicky pavilon, Slovenska vysoka skola technicka); SAJTER, Vit, inz. (Bratislava, Sasinkova 4)

Automatic radiometric titration to the point of equivalence. Chem zvesti 16 no.3:217-223 Mr '62

- 1. Katedra radiochemie a radiacnej chemie Slovenskej vysokej skoly technickej, Bratislava (for Tolgyessy)
- 2. Katedra biochemie Lekarskej fakulty Univerzity Komenskeho, Bratislava (for Sajter).



L	1634-66 EWP(t)/EWP(b) DIAAP/IJP(c) JD/JG
	ACCESSION NR: AP5024262 AUTHOR: Stefenov. G. (Sofia); Nenov. H. (Sofia); Tomov. T. (Sofia); Zivkov. Social; Chivkov, Zh.)(Sofia); Georgiev. N. (Georgiyev, N.)(Sofia); Popov. C. (Popov. Ra.) (Sofia); Hichailov. H. (Hikhaylov, H.)(Sofia); Tolgressy, J. (Tel'deehi, Tu.) (Engineer, Docent, Candidate of sciences) (Bratislava) TITLE: Determination of gold in mineral raw materials by means of the neutron activation analysis SOURCE: Chemicke svesti, no. 9, 1964, 661-668
	TOPIC TAGS: gold, analytic chemistry, silicate, radiation spectrometer, radiometer, radiation chemistry, neutron irradiation, neutron flux, neutron Abstract [Authors' German summary, modified]: A method is presented of determining gold in samples of silicates by means of the neutron activation method. Samples were irradiated in a nuclear reactor by a flux of neutrons of 2-li by 10 ¹² n by cm ² by s ² . The induced activity was measured by a 400% channel scirtillation \(\gamma\) spectrometer or a B-2 radiometer. It is possible to determine gold in ore and non-ore raw materials up to the volume of 2.8 by 10 ¹⁰ gramma*. Origo art. has 2 graphs and 3 tables.
	Cert 1/2
•	

	1634-66 ACCESSION NR: AP5024262
	ASSOCIATION: Stefanov, Nenov, Tomov, Zivkov, Georgiev, Popov, Michajloy Naucno izeledovatelski geologiceski institut pri Glavno upravlenie po geologija i ochrano na zemnite nedra, laboratorja aktivacionen analiz, Sofia (Activation Analysis
	Institute and a laboratory and traction of Geology, Main Administration for the Geology and Protection of Mineral Resources (5) Tolgressy Katedra radiochemie a radiacnej chemie Slovenskej vysokej skoly technickej, Bratislava (Slovek Institute of Technology, Department of Endiochemistry and Endiation Chemistry)
	SUBMITTED: 23Jam64 EXCL: 00 SUB CODE: QC, NP NO REF SOV: 004 OTHER: 007 JPRS
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EWP(j)/T AT6028246 ACC NR: SOURCE CODE: HU/2502/65/046/001/0035/0044 AUTHOR: Braun, Tibor (Doctor; Budapest); Hradil, M.—Khradil, M. (Bratislava); Jesenak, V.—Yesenak, V. (Bratislava); Tolgyessy, J.—Tel'deshi, Y. (Doctor; Bratislava) ORG: [Braun] Institute of Inorganic and Analytical Chemistry, L. Ectvos University, Budapest; [Hradil; Jesenak; Tolgyessy] Department of Radiochemistry and Radiation Chemistry, Faculty of Chemistry, Slovak Technical University, Bratislava, Czechoslovakia TITIE: Radiocoulometric titrations SOURCE: Academia scientiarum hungaricae. Acta chimica, v. 46, no. 1, 1965, 35-44 TOPIC TAGS: titrimetry, radiation chemistry, radioisotope ABSTRACT: Two methods, one intermittent and the other continuous, have been developed for radiometric determination of the end point of coulometric titrations based on formation of precipitate and complexation. In the radiocoulometric titrations based on precipitate formation, iodide ions labeled with Il31 were titrated with silver ions generated by coulometry, using silver electrodes. In the complexometric radiocoulometric titrations with the aid of a solid indicator, the cyanide ions generated by the electrolysis of the complex [Ag(CN)2] were reacted with the Ni ions to be determined, using AgI solution containing labeled Ag. The experimental apparatus is described. Orig. art. has: 5 figures and 1 table. [Orig. art. in Eng.] [JPRS: 33,906] SUB CODE: 07 / SUBM DATE: 30Jan65 / ORIG REF: 010 / OTH REF: 002

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L 37249-66 EWI(m)		20
ACC NR: AP6027865	SOURCE CODE: CZ/0038/66/000/	003 (0081 (0088
AUTHOR: Tolgyessy, Juraj Tel'deshi, Yu.		34 B
ORG: Department of Radiochemistry and Ra Faculty, Slovak Institute of Technology, chemie, Chemicko-technologicka fakulta, S		ineering
TITIE: Neutron absorption analysis		
SOURCE: Jaderna energie, no. 3, 1966, 81	-8 8	
TOPIC TAGS: neutron absorption, particle	physics, physics laboratory ins	strument
ABSTRACT: This survey article presents to of instruments for neutron absorption and with consideration of its advantages and data available to the author. This paper has: 8 figures, 6 formulas and 1 table.	he principles and methods and de lysis as well as its practical a disadvantages on the basis of th	escriptions applications
	IG REF: 007 / SOV REF: 030	
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Card 1/1	UDC: 543.53	
	0917	1378

1 410/5-66 EMP(1) IJP(c) MM
SOURCE CODE: CZ/0043/66/000/003/0206/0208
AUTHOR: Dillinger, Pavol (Graduate chemist); Tolgyessy, Juraj-Tel'deshi, Yu. (Docent; Engineer; Candidate of sciences)
ORG: Department of Radiochemistry and Radiation Chemistry, SVST, Bratislava (Katedra radiochemie a radiacnej chemie SVST)
TITIE: Simple scintillation measuring head for the determination of the absorption of beta radiation in liquid samples
SOURCE: Chemicke zvesti, no. 3, 1966, 206-208
TOPIC TAGS: radiation chemistry, scintillator
ABSTRACT: The authors describe an apparatus of their design. One of the walls of the through-flow cell is the beta-scintillator, and the other consists of a lmm thick plexiglass plate. The apparatus is suitable for batch analysis of binary liquid mixtures, for radiometric beta absorption titrations, and can also be adapted for continuous analysis. Orig. art. has: 2 figures. [JPRS: 36,002]
SUB CODE: 07, 18 / SUBM DATE: 18Jul65 / ORIG REF: 004 / OTH REF: 002
Card 1/1 hs
0918 2634

HUNGARY / Chemical Technology, Chemical Products and Their H-34 Application. Dying and Chemical Treatment of Textile Materials.

: Ref Zhur - Khimiya, No 5, 1959, No. 17900 Abs Jour

: Rusznak, I.; Fehervari, M.; Tolgyesi, L.; Ban, G. Author

Inst : Not given

Title : Decomposition of o-Aminoazobenzene Under the Action of

Ultraviolet Rays

Orig Pub : Magyar textiltechn., 1957, No 2, 73-74

: Comparison of the rate of decomposition of o- and n-Abstract

aminoazobenzene under the action of ultraviolet rays and chemical oxidation indicates that the ortho-derivative under all conditions proves to be relatively more stable. Evidently, the amino-group when in ortho position protects

the azo-group. -- G. Yudkovich

Card 1/1

H-155

SARSUNOVA, Magda, dr., FhMr., CSc.; KLAS, Jan, promovany chemik; TOLGYESSY, Juraj, doc., inz., CSc.

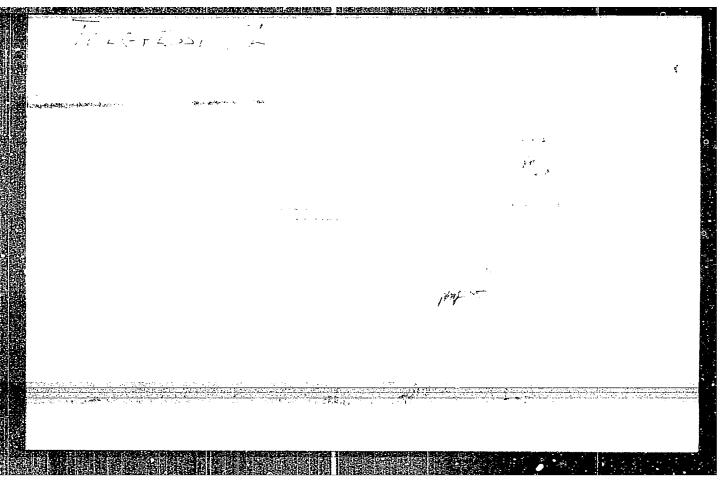
Determining copper and zinc in some drugs by the measurment of intensity of reflected beta rays. Chem zvesti 17 no.7:504-509 163.

1. Krajske kontrolne laboratorium, Krajsky ustav narodneho zdravie Zapadoslovenskeho kraja, Bratislava, Radlinskeho 20 (for Sarsunova). 2. Katedra radiochemie a radiacnej chemie, Slovenska vysoka skola technicka, Bratislava, Kollarovo namesti 2 (for Klas, Tolgyessy).

TOLGYESSY, J.

An account of the 3d nationwide conference on radiochemistry. Chem zvesti 19 no.2:151-152 '65.

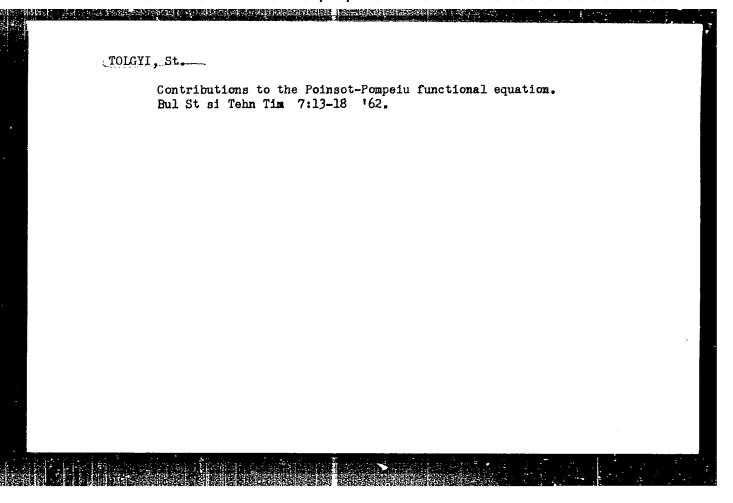
1. Editorial Board Member, "Chemicke zvesti."

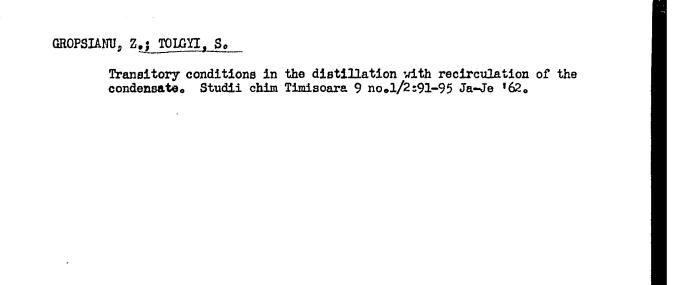


TOLGYI, S.; GROPSIANU, Z.

The residence of the control of the

On a cyclic system of linear differential equations. Studii chim Timisoara 10 no.1:115-124 Ja-Je '63.





TOLIBEKOV, D.

Photochemical reaction of the transformation of xanthophylls in a reconstructed system. Bot, zhur. 48 no.9:1383-1384 S '63. (MIRA 16:11)

1. Botanicheskiy institut imeni V.L. Komarova AN SSSR, Leningrad.

Carrying out the light conversion reaction of xanthophylls in reproduced systems. Dokl. AN Tadzh. SSR 6 no.4:37-41 163.

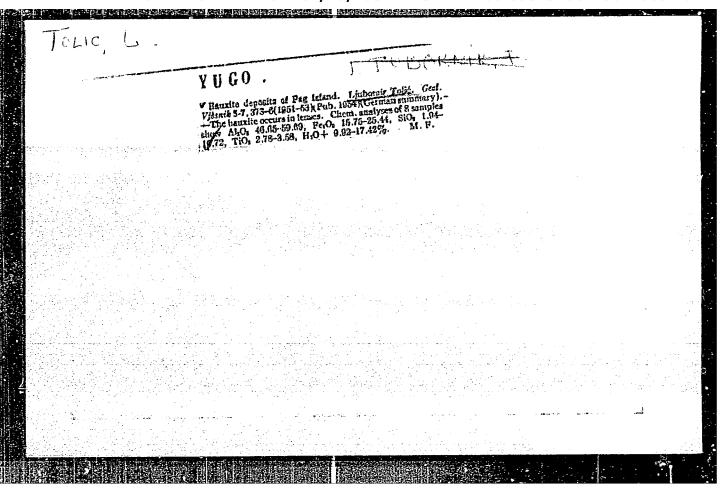
(MIRA 17:4)

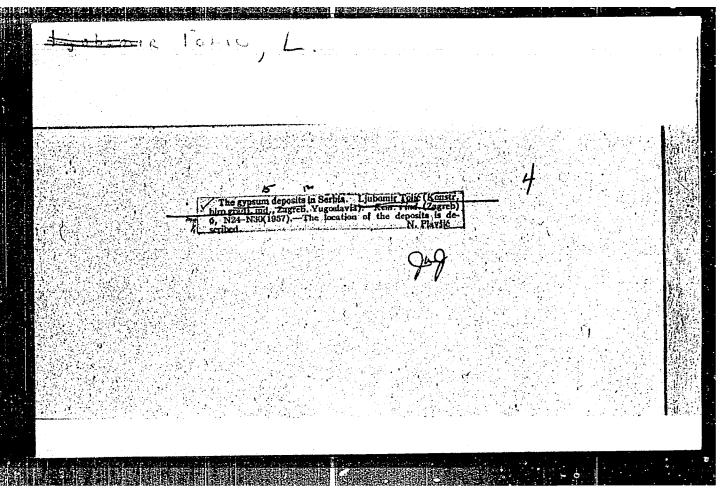
1. Otdel fiziologii i biofiziki rasteniy AN Tadzhikskoy SSR. Predstavleno akademikom AN Tadzhikskoy SSR K.T.Poroshinym.

SAPOZHNIKOV, D.I.; EYDEL'MAN, Z.M.; TOLIBEROV, D.; KHODZHAYEV, A.

Transformation of xanthophylls in partially reconstructed systems. Bot.zhur. 47 no.11:1656-1659 N '62. (MIRA 16:1)

1. Botanicheskiy institut imeni V.L.Komarova AN SSSR, Leningrad. (Photosynthesis) (Xanthophylls)





APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756110002-8"

TOLIC, L.

YUGOSLAVIA / Cosmochemistry, Geochemistry, Hydro- D chemistry.

Abs Jour: Ref Zhur-Khimiya, 1958, No 20, 67208.

Author : Tolic L. Inst : Not given.

Title : Deposits of Quartz Sand in Vranye, Near Biograd-

on-the-Sea (Dalmacija).

Orig Pub: Kemij u industriji, 1957, 6, No 4, 2-3.

Abstract: A stratum of quartz sand laminated in limestone layers was found in the above location. This sand is basically quartz but it contains small quantities of muscovite (mica) and apatite (asparagus stone). Chemical composition of the raw material and of it's 0.004-0.04 mm fraction (respectively in %): SiO₂-93.34, 94.20; Fe₂O₃-0.98, 0.84; Al₂O₃-3.00, 2.57; CaO-0.34, 0.24; Mg O-0.18, 0.10; MnO-

Card 1/2

18

YUGOSLAVIA / Cosmochemistry. Geochemistry. Hydro-Dehemistry.

Abs Jour: Ref Zhur-Khimiya, 1958, No 20, 67208.

Abstract: traces, traces; H_2O (-110°)-0.51, 0.33; H_2O (+110°) -1.14, 1.20; Total- 99.49, 99.48.

Card 2/2

TOLIC, L.

CZECHOSLOVAKIA / Cosmochemistry. Geochemistry. Hydro- D chemistry.

Abs Jour: Ref Zhur-Khimiya, 1958, No 20, 67211.

Author : Tolic L.
Inst : Not given.

Title : Deposits of Gypsum in the Macedonian Peoples Re-

public.

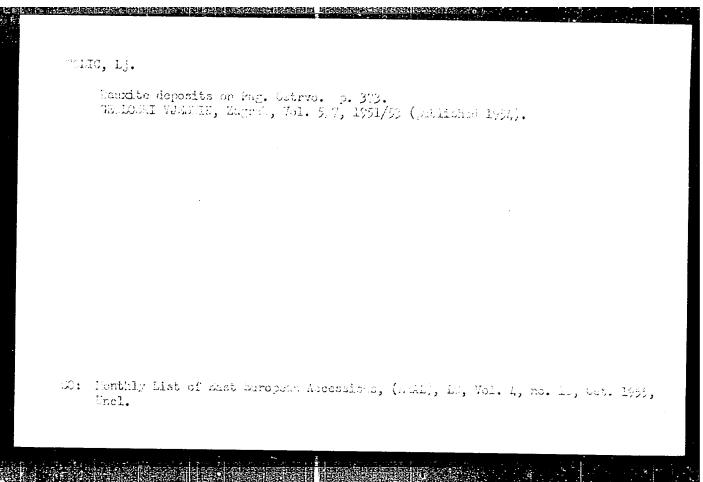
Orig Pub: Kemija u industriji, 1957, 6, No 4, 8-11.

Abstract: The deposits of high quality gypsum with the total potential of 66 millions tons were discovered in the region of Kosovrast. Gypsum and alabaster layers are intermixed with pools of crystalline gypsum. They rest on the stratum of clay formation. A map showing geological sections and chemical an-

alyses is given.

Card 1/1

19



TOLIC, Liubomir, prof.

Gypsum deposits in Bosnia and Hercegovina. Kem ird 13 no. 2: 85-91 F '64.

Gypsum deposits in Slovenia. Ibid.: 91-92.



JOVIC, Mihailo, inz.; MOJKOVIC, Dusan, inz.; TOLIC, Natalija, tehm.

Electronic track circuits. Zeleznice Jug 19 no.5:38-47 My*63.

FEDCHIN, V.F.; TOLIKINA, N.F.; BELYATSKAYA, O.N.; GUL', V.Ye.

CHARLES EN FORTH MAINTENANT SERVICE BETWEEN SERVICES STREET, TOTAL SERVICES OF THE SERVICES S

Composition of impurities in straight-chain paraffinio hydrolarkons having analytical application. Zhur. anal. khim. 20 nc.9: 1022-1024 165. (MIRA 1889)

l. Moskovskiy tekhnologicheskiy institut myasacy i molechnoy promyshlennosti.

TOLIMANCHUK, L.F.; TYUTIN, A.A.

Inverse function generator using diode elements. Izv. vys. ucheb. zav.; radiotekh. 6 no.1:24-32 Ja-F '63. (MIRA 16'3)

1. Rekomendovana kafedroy vychislitel'noy tekhniki Kiyevskogo ordena Lenina politekhnicheskogo instituta. (Oscillators, Transistor) (Pulse techniques (Electronics))

ACCESSION NR: AP4020323

\$/0302/64/000/001/0065/0066

AUTHOR: Levchenko. N. A.; Tolimanchuk, V. A.

TITLE: Simple pulse counter

SOURCE: Avtomatika i priborostroyeniye, no. 1, 1964, 65-66

TOPIC TAGS: pulse counter, transistorized pulse counter, ferrite pulse counter,

ABSTRACT: Ferrite-core windings (see Enclosure 1) are so connected that the current charging C_{κ} brings the ferrite to its zero state, i.e., records a binary zero. The discharge current through R_{κ} transfers the core to the unit state and the preceding core to the zero state, and so on. A 10-digit counter hookup with stable operation with the power-supply and bias voltages varying within \pm 20%. Orig. art. has: 2 figures.

ASSOCIATION: none

SUBMITTED: 00

DATE ACQ: 31Mar64

ENCL: 01

SUB CODE: CP, GE

NO REF SOV: 000

OTHER: 000

Card 1/2/

TOLIA, V

SUBJECT: USSR/Bibliography

4-4-3/22

AUTHOR:

Vzorov, M., Gurevich, N., Tolin, V., Gritchuk, A.

TITLE:

New Publications (Vyshli iz pechati)

Particular de la companya del companya del companya de la companya

PERIODICAL:

Znaniye - Sila, April 1957, #4, pp 3-4 (USSR)

ABSTRACT:

The four authors review 4 books all dealing with reminiscences of Lenin. The author of the first book "Reminiscences of V.I. Lenin" is not indicated. A. Bezymenskiy is the author of the second book entitled "Encounters of Komsomol'tsy with V.I. Lenin" (Vstrechi komsomol'tsev s V.I. Leninym). The third book is

(Vstrechi komsomol'tsev s V.I. Leninym). The third book is written by V. Bonch-Bruyevich "V.I. Lenin in Petrograd and Moskva (1917-1920)" and the fourth one "Lenin's Youth" by N. Nechvolo-

dova and L. Reznichenko.

The article contains 4 photos of the book covers.

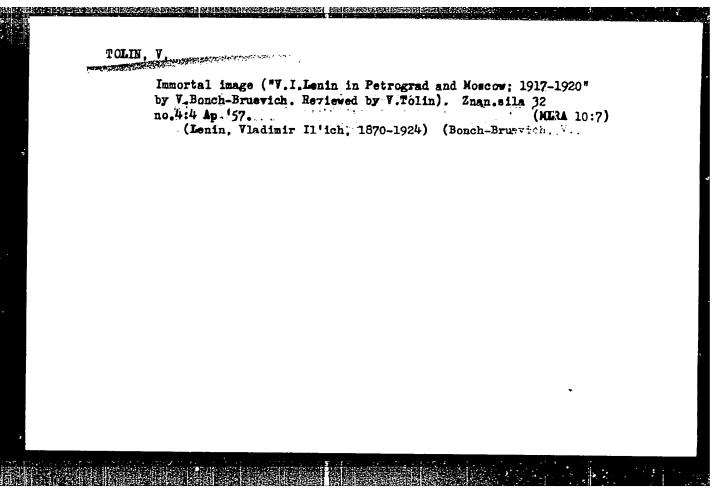
ASSOCIATION: -

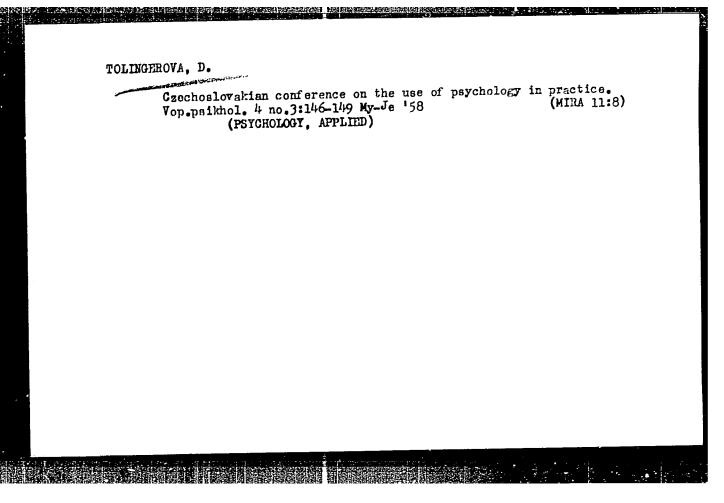
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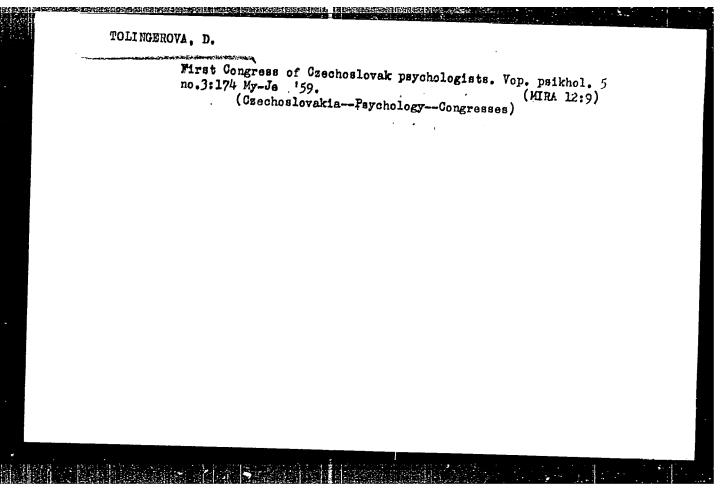
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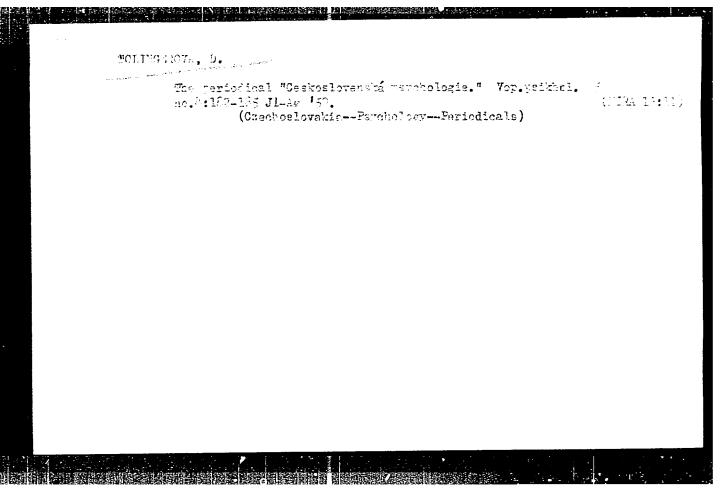




TOLINGEROVA, D. Cand Ped Sci -- (diss) "Georgiy Prokhazka and His Place in the History of Psychology." Mos, 1957. 19 pp 20 cm. (Mos Order of Lenin and Order of Labor Red Banner State Univ im M. V. Lomonosov), 100 copies (KL, 25-57, 120)

- 15





YUGOSLATIA / General and Special Zoology. Insects. P Harmful Insects and Mites. Posts of Commercial, Oil-Bearing, Medicinal and Essen-

tial Oil-Boaring Crops.

Abs Jour: Ref Zhur-Biol., No 1, 1959, 2314.

Author : Tolja, J. Inst : Not given.

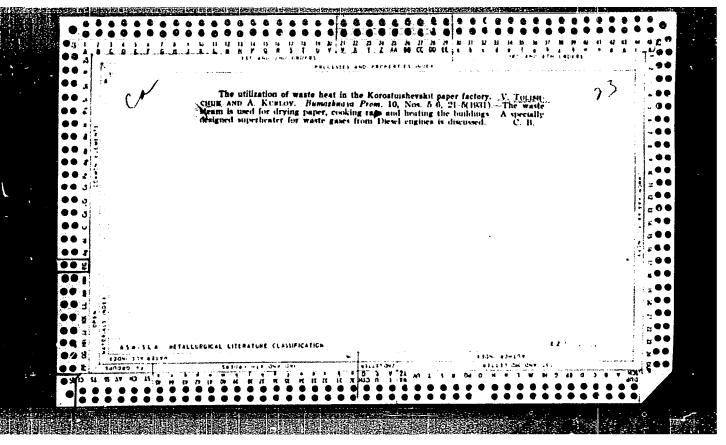
Title : Two Dangerous Tobacco Pests.

Orig Pub: Tutun, 1957, 7, No 12, 454-459.

Abstract: No abstract.

Card 1/1

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TOLJANIC, D.

Tony, the road-maintenance man. p. 239
CESTE I MOSTOVI, Zagreb, Vol 4, No. 6, June, 1956

SO: East European Accessions List, Vol 5, No. 10, Oct, 1956

TOLIMITO, D.

T'LUANIC, D. What does the roadman Ljube thinks? p. 500.

Vol. L, No. 11, Nov. 1956. CESTE I MOSTOVI TECHNOLOGY Zagreb, Kugoslavia

So: East European Accession, Vol. 6, No. 2, February 1957

BATINIC, Nikola, ing. (Beograd, Svetog Save 32); TOLJEVIC, Radmila, ing. (Beograd, Svetozara Markovica 5)

Analysis of the hydraulic calculation of Sava River at Brcko. Brodarstvo 4 no. 13:549-558 O-D 61.

SOSHIN, 1.0.; TCLKatil, cod.

Low-noise preamplifier for spectromotors equipped with semiconductor transducers. brib. i takh. eksp. 8 nc.6:127-126

N-D '63. (Mink 17:6)

1. Belorusskiy gosudaratvennyy universitet.

TOL	KACHENKO, A.
	Reducing the number of crushed grains in corn shelling. Muk-elev. prom. 24 no.6:27 Je '58. (MIRA 11:7)
	1. Odesskaya realizatsionnaya baza khleboproduktov. (Corn (Maize))

TOLKACHEV, A., inzh.

Radioactive indicator. Muk. elev. prom. 24 no.26-27 ll '58.
(MIRA 11:12)

1.Tekhnicheskiy otdel Ministerstva khleboproduktov SSSR.
(Radioactive tracers)
(Flour mills--Equipment and supplies)

SOV/112-59-1-1446

Translation from: Referativnyy zhurnal. Elektrotekhnika, 1959, Nr 1, p 198 (USSR)

AUTHOR: Tolkachev, A.

TITLE: Electronic Machines and Problems of Planning

PERIODICAL: Prom.-ekon. gaz., March 19, 1958, Nr 34, p 2

ABSTRAGT: A report on projects done at the Nauchno-issledovatel'skiy ekonomicheskiy institut Gosplana SSSR (Scientific-Research Economic Institute) on the possibilities of using electronic computers in planning and economic estimates.

Card 1/1

TOLKACHEV, A., inzh.

Formwork for mamufacturing reinforced corcrete arches. Sel'.
stroi. l6 nc.9:Insert; 2 S '61. (MIRA 14:9)

(Arches)

(Reinforced concrete construction—Formwork)

ALEKSEYEV, A.; ANCHISHKIN, A.; BERRI, L.; BARABANOV, M.; BOGOMOLOV, O.;
BRAGINSKIY, B.; IOFFE, Ya.; KOVAL', T.; KONAKOV, D.; KUVARIH, V.;
KUUROV, V.; LITVYAKOV, P., NURONTSEV, M.; OBOLENSKIY, K.; POKATAYEV,
Yu.; TOLKACHEV, A.; KATS, V., red.; KRYLOV, P., red.; KANEVSKAYA,
T.M., red.; GERASINOVA, Ye.S., tekhn.red.

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[Economic competition between the U.S.S.R. and the U.S.A.; a criticism of the views of American bourgeois economists] Ekonomicheskoe sorevnovanie mezhdu SSSR i SShA; kritika vzgliadov amerikanskikh burshuasnykh ekonomistov. Moskya, Gosplanizdat, 1959. 240 p. (MIRA 12:3)

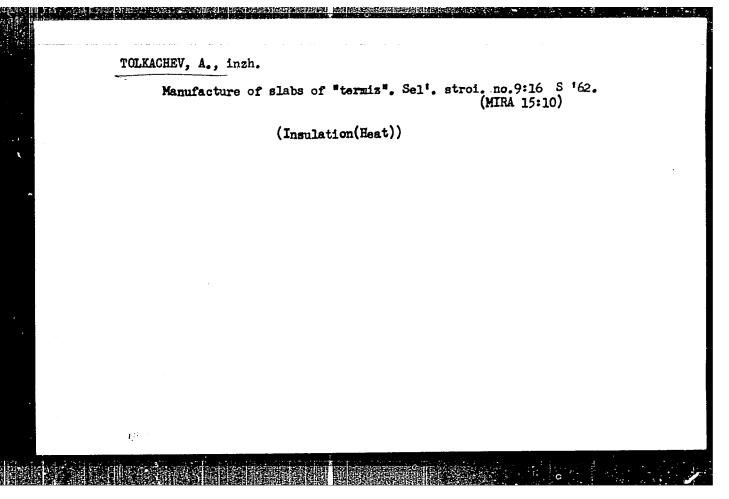
1. Moscow. Mauchno-issledovatel'skiy ekonomicheskiy institut. 2. Sotrudniki Mauchno-issledovatel'skogo ekonomicheskogo instituta Gosplana SSSR
(for all except Kats, Krylov, Kanevskaya, Gerasimova)
(United States--Economic conditions) (Russia--Economic conditions)

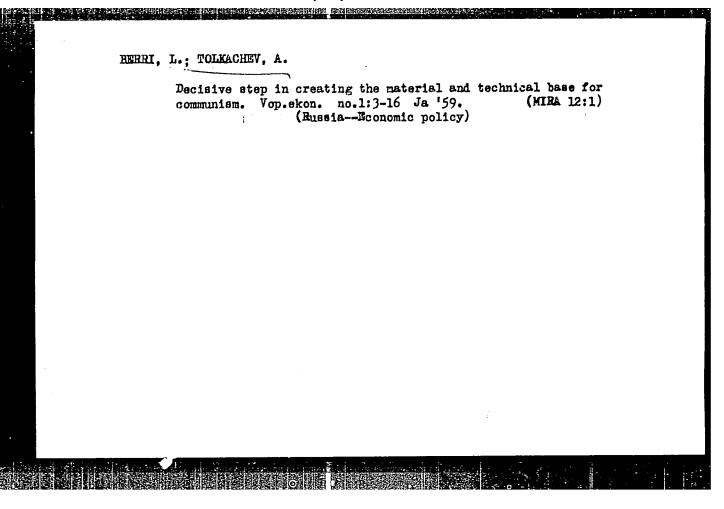
TOLKACHEV, A.

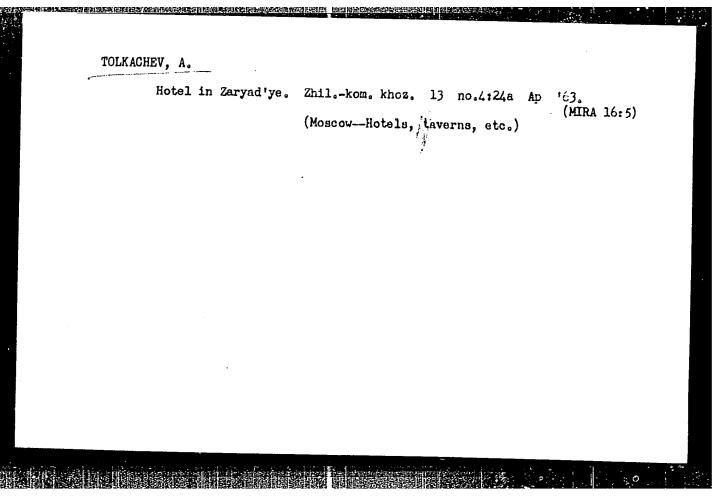
Twenty fiery minutes. Pozh.delo 7 nc.12:16-17 D '61. (MIRA 14:11)

1. Glavnyy inzh. Tuapsinskogo neftepererabatyvayushchego zavoda, presedatel pozharno-tekhnicheskoy komissii zavoda, g. Tuapse. (Tuapse--Petroleum industry--Fires and fire prevention)

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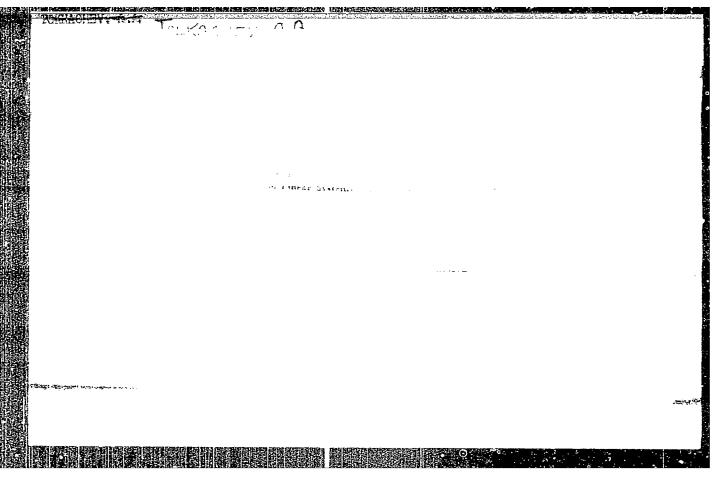


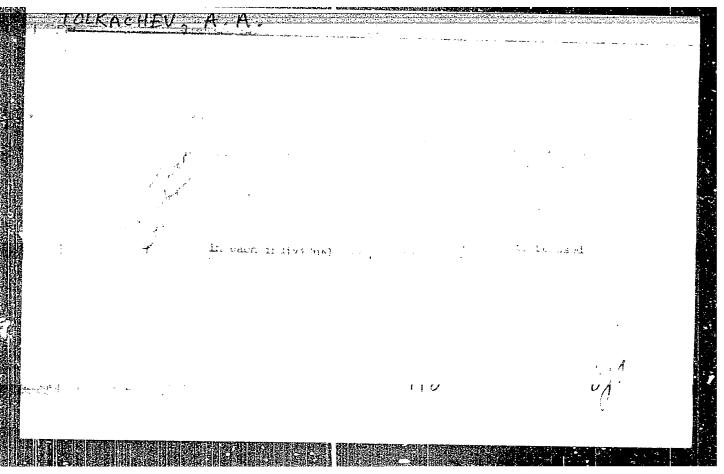


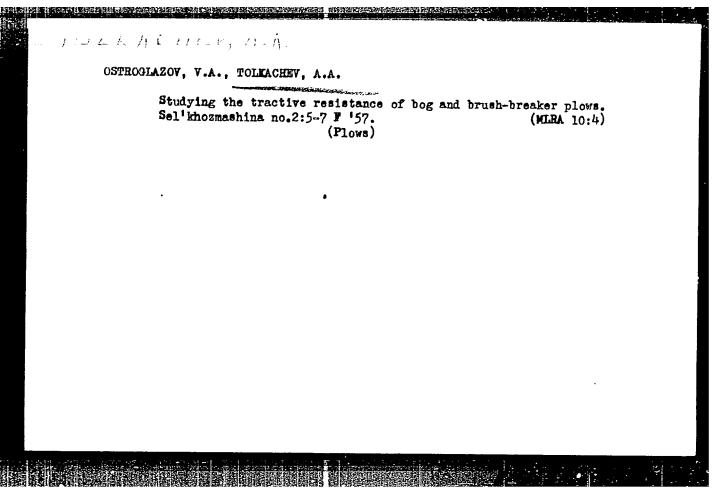
BERRI, Lev Yakovlevich; TOLKACHEV, Aleksandr Sergeyevich; PISKUNOV, V., red.; DANILINA, A., tekhn.red.

[Economic and technicel foundation of communism] Material'notekhnicheskaie baza kommunizma. Moskva, Gos.izd-vo polit.lit-ry, 1960. 97 p. (MIRA 13:7)

(Russia--Industries) (Communism)







BARABANENKOV, Yu.N.; TOLKACHEV, A.A.; AYTKHOZHIN, N.A.; LESOTA, O.K.

Scattering of an electromagnetic 8-impulse on perfectly conducting bodies with finite dimensions. Radiotekh. i elektron. 8 no.6: (MIRA 16:7)

1069-1071 Je '63. (Electromagnetic waves)

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